organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Jun Wan, Zheng-Zhong Peng, Xue-Mei Li and Shu-Sheng Zhang*

College of Chemistry and Molecular Engineering, Qingdao University of Science and Technology, 266042 Qingdao, Shandong, People's Republic of China

Correspondence e-mail: shushzhang@126.com

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.003 Å R factor = 0.046 wR factor = 0.116 Data-to-parameter ratio = 14.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(1*H*-1,2,3-Benzotriazol-1-ylmethyl)-1-benzoylethyl 4-chlorobenzoate

In the title compound, $C_{22}H_{16}ClN_3O_3$, molecules are linked into chains along the *c* axis by $C-H\cdots O$ hydrogen bonds, while other $C-H\cdots N$ hydrogen bonds connect the chains into two-dimensional layers. The packing is further stabilized by $\pi-\pi$ interactions.

Comment

Benzotriazole derivatives have been found to exhibit a broad spectrum of pharmacological activities such as anti-inflammatory, antifungal, antitumor, antineoplastic and anti-depressant activities (Al-Soud *et al.*, 2003; Katarzyna *et al.*, 2004). As part of a search for new benzotriazole compounds with high bioactivity, the title compound, (I), was synthesized.



The bond lengths and angles in (I) show normal values (Allen *et al.*, 1987) and are comparable to those in the related compound 2-(1*H*-1,2,3-benzotriazol-1-ylmethyl)-1-benzoylethyl 4-ethylbenzoate (Wan *et al.*, 2005). The benzotriazole system is essentially planar, with a dihedral angle of 1.9 (1)° between the triazole ring (*A*, atoms N1–N3/C10/C15) and the benzene ring (*B*, atoms C10–C15). The dihedral angles between the mean planes of the benzotriazole system and rings *C* (atoms C1–C6) and *D* (atoms C17–C22) are 38.1 (1) and 53.7 (1)°, respectively. The dihedral angle between rings *C* and *D* is 88.9 (1)°.

In the crystal structure, molecules of (I) are linked into chains along the *c* axis by C9–H9A···O1ⁱⁱ and C11– H11A···O1ⁱⁱ hydrogen bonds (see Table 2 for geometry values and symmetry codes). In addition, C3–H3B···N3ⁱ interactions connect the chains into two-dimensional layers in the *ac* plane (Fig. 2). The packing is further stabilized by π – π interactions involving the benzotriazole system and benzene ring *D*, the distances being Cg1··· $Cg4^{iii} = 3.570$ Å and Cg3··· $Cg4^{iii} = 3.654$ Å [Cg1, Cg3 and Cg4 denote the centroids of the rings *A*, *B* and *D*, respectively; symmetry code: (iii) $x, \frac{3}{2}$ – $y, z - \frac{1}{2}$]. Received 5 January 2006 Accepted 11 January 2006

© 2006 International Union of Crystallography

All rights reserved



Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme.

Experimental

Bromine (3.2 g, 0.02 mol) was added dropwise to a solution of 3-(benzotriazol-1-yl)-1-phenylpropan-1-one (5.0 g, 0.02 mol) and sodium acetate (1.6 g, 0.02 mol) in acetic acid (50 ml). The reaction was allowed to proceed for 13 h. Water (50 ml) and chloroform (20 ml) were then added. The organic layer was washed successively with saturated sodium bicarbonate solution and brine, dried over anhydrous magnesium sulfate and the chloroform solution filtered. It was cooled with ice-water and then an acetone solution (10 ml) of 4chlorobenzoic acid (3.1 g, 0.02 mol) and triethylamine (2.8 ml) were added. The mixture was stirred at room temperature for about 2 h. The solution was then filtered and concentrated (yield 62%). Crystals of (I) were obtained by slow evaporation of an acetone–water (1:1 v/v) solution at room temperature over a period of one week.

Crystal data

	2		
$C_{22}H_{16}CIN_3O_3$	$D_x = 1.361 \text{ Mg m}^{-3}$		
$M_r = 405.83$	Mo $K\alpha$ radiation		
Monoclinic, $P2_1/c$	Cell parameters from 2380		
a = 9.712 (3) Å	reflections		
b = 20.379 (5) Å	$\theta = 2.3-22.3^{\circ}$		
c = 10.009 (3) Å	$\mu = 0.22 \text{ mm}^{-1}$		
$\beta = 90.392 \ (4)^{\circ}$	T = 293 (2) K		
$V = 1980.9 (9) \text{ Å}^3$	Column, colorless		
Z = 4	$0.28 \times 0.11 \times 0.09 \ \mathrm{mm}$		
Data collection			
	2005 1 1 4 9 4		

Siemens SwiART 1000 CCD area-
detector diffractometer
ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\min} = 0.941, T_{\max} = 0.967$
11133 measured reflections

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.046$
$wR(F^2) = 0.116$
S = 1.02
3885 reflections
262 parameters
H-atom parameters constrained

Joos independent reflections
2719 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.027$
$\theta_{\rm max} = 26.0^{\circ}$
$h = -11 \rightarrow 10$
$k = -25 \rightarrow 19$
$l = -12 \rightarrow 12$

 $w = 1/[\sigma^2(F_0^2) + (0.0474P)^2]$ + 0.3694P] where $P = (F_0^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.25 \ {\rm e} \ {\rm \AA}$ $\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$



Figure 2

The two-dimensional layers of (I), viewed down the b axis. Dashed lines indicate C-H···O.N interactions.

Table 1 Selected bond lengths (Å).

Cl1-C20	1.743 (2)	O3-C16	1.202 (2)
O1-C7	1.218 (2)	C7-C8	1.527 (3)
O2-C16	1.357 (2)	C8-C9	1.514 (3)
O2-C8	1.436 (2)		

Table 2

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C3 - H3B \cdots N3^{i}$	0.93	2.58	3.506 (4) 3.453 (3)	175 173
$C11 - H11A \cdots O1^{ii}$	0.97	2.49	3.316 (3)	143

Symmetry codes: (i) x + 1, $-y + \frac{3}{2}$, $z - \frac{1}{2}$; (ii) $x, -y + \frac{3}{2}$, $z - \frac{1}{2}$.

All H atoms were located in difference Fourier maps and constrained to ride on their parent atoms, with C-H distances in the range 0.93–0.98 Å and with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

This study was supported by the Project of Educational Administration of Shandong Province (No. J04B12) and the Outstanding Adult-Young Scientific Research Encouraging Foundation of Shandong Province (No. 2005BS04007).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Al-Soud, Y. A., Al-Masoudi, N. A. & Ferwanah, A. S. (2003). Bioorg. Med. Chem. 11, 1701-1708.

Katarzyna, K., Najda, A., Justyna, Z., Chomicz, L., Piekarczyk, J., Myjak, P. & Bretner, M. (2004). Bioorg. Med. Chem. 12, 2617-2624.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

- Sheldrick, G. M. (1997). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). SMART and SAINT. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Spek, A. L. (2003). J. Appl. Cryst. **36**, 7–13. Wan, J., Peng, Z.-Z., Li, X.-M. & Zhang, S.-S. (2005). Acta Cryst. E**61**, 04218– o4219.